

# Think of coordinate rescaling

$$\vec{x} = \alpha \vec{\mu}$$

$$\vec{x} = (x_1, x_2, x_3) \quad , \quad \vec{\mu} = (\mu_1, \mu_2, \mu_3)$$

$$\vec{\nabla}_x^2 = \frac{\partial}{\partial x_1^2} + \frac{\partial}{\partial x_2^2} + \frac{\partial}{\partial x_3^2} \quad , \quad \vec{\nabla}_\mu^2 = \frac{\partial}{\partial \mu_1^2} + \frac{\partial}{\partial \mu_2^2} + \frac{\partial}{\partial \mu_3^2} = \alpha^2 \left( \frac{\partial}{\partial x_1^2} + \frac{\partial}{\partial x_2^2} + \frac{\partial}{\partial x_3^2} \right)$$

$$\vec{x} = \alpha \vec{\mu} \quad , \quad \partial_x^2 = \frac{\partial^2}{\partial \vec{x}^2} = \frac{1}{\alpha^2} \partial_\mu^2 = \frac{1}{\alpha^2} \frac{\partial^2}{\partial \vec{\mu}^2}$$

# For example, the Schrodinger of H atom

$$\left( -\frac{\hbar^2}{2m} \nabla^2 - \frac{e^2}{r} \right) \psi_n(\vec{r}) = \varepsilon_n \psi_n(\vec{r})$$

Introduce a dimension-less length  $\vec{y}$

$$\vec{r} = \alpha \vec{y} \quad \text{and then we have} \quad \frac{\partial}{\partial \vec{r}} = \frac{1}{\alpha} \frac{\partial}{\partial \vec{y}}, \quad \vec{\nabla}^2 = \frac{\partial^2}{\partial \vec{r}^2} = \frac{1}{\alpha^2} \frac{\partial^2}{\partial \vec{y}^2}$$

$$\left( -\frac{\hbar^2}{2m} \frac{1}{\alpha^2} \nabla_y^2 - \frac{e^2}{\alpha y} \right) \psi_n(\alpha \vec{y}) = \varepsilon_n \psi_n(\alpha \vec{y})$$

$$\left( -\frac{1}{2} \nabla_y^2 - \left( \frac{m}{\hbar^2} \alpha e^2 \right) \frac{1}{y} \right) \varphi_n(\vec{y}) = \frac{\varepsilon_n}{\left( \frac{\hbar^2}{m\alpha^2} \right)} \varphi_n(\vec{y})$$

Let us take  $\alpha = \frac{\hbar^2}{me^2} = a_0$  : Bohr radius.

And then we have  $\frac{\hbar^2}{m\alpha^2} = \frac{me^4}{\hbar^2} = \frac{e^2}{a_0} = H_a$

# Now we have the equation in dimension-less number

$$\left( -\frac{1}{2} \nabla^2 - \frac{1}{y} \right) \psi_n(\vec{y}) = E_n \psi_n(\vec{y})$$

- In this equation, the length is in the unit of  $a_0$ .
- Note that  $a_0 = 0.52946541 \text{ \AA}$ .
- The energy is in unit of  $H_a$ . Note that  $H_a = 4.35974e^{-18} \text{ Joule}$ .
- Note that  $Ry = \frac{e^2}{2a_0} = \frac{1}{2} H_a = 13.6 eV$

# Atomic-Hartree systems of unit

The equation, Kohn-Sham DFT equation

$$\left( -\frac{\hbar^2}{2m} \nabla^2 + U_{atom}(\vec{r}) + \int \frac{e^2 \rho(\vec{r}')}{|\vec{r} - \vec{r}'|} d^3 \vec{r}' + v_{xc}[\rho(\vec{r})] \right) \psi_n(\vec{r}) = \varepsilon_n \psi_n(\vec{r})$$

By setting  $m = 1, e = 1, \hbar = 1$ , we have the atomic-Hartree systems of unit.

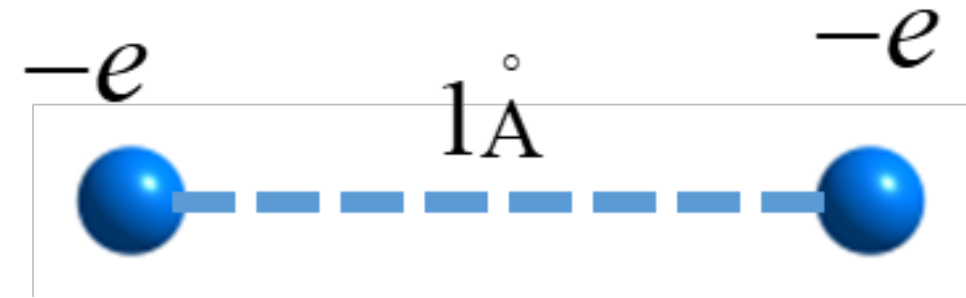
$$a_0 = \frac{\hbar^2}{me^2} = 1, \quad R = \frac{e^2}{2a_0} = \frac{me^4}{2\hbar^2} = \frac{1}{2}, \quad H_a = \frac{e^2}{a_0} = \frac{me^4}{\hbar^2} = 1$$

# [Example] Atomic-Hartree systems of unit

Calculate the potential energy between two point charges with the charge  $-e$ , separated by  $1 \text{ \AA}$

Distance = (1 angstrom) =  $1/0.529$

$$E = \frac{q_1 q_2}{r} = \frac{1}{(1/0.529)} = 0.529$$



# [Example] Atomic-Hartree systems of unit

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Calculate the potential energy between two point charges with the charge  $-e$ , separated by  $1 \text{ \AA}$

$$E = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r} = 8.998 \times 10^9 \left[ \frac{Nm^2}{C^2} \right] \times \frac{(1.6 \times 10^{-19})^2 [C]^2}{10^{-10} [m]} \approx 2.3035 \times 10^{-18} [J]$$

$$0.529 [H_a] = 2.3035 \times 10^{-18} [J] \quad , \quad 1 [H_a] \approx 4.3544 \times 10^{-18} [J]$$

# Atomic-Hartree systems of unit

The equation, Kohn-Sham DFT equation

$$\left( -\frac{1}{2} \nabla^2 + U_{atom}(\vec{r}) + \int \frac{\rho(\vec{r}')}{|\vec{r} - \vec{r}'|} d^3\vec{r}' + v_{xc}[\rho(\vec{r})] \right) \psi_n(\vec{r}) = \varepsilon_n \psi_n(\vec{r})$$

By setting  $m = 1, e = 1, \hbar = 1$ , we have the atomic-Hartree systems of unit.



# Atomic-Rydberg systems of unit

1. Atomic Rydberg systems of unit can be derived by setting

$$a_0 = \frac{\hbar^2}{me^2} = 1, \quad R = \frac{e^2}{2a_0} = \frac{me^4}{2\hbar^2} = \frac{1}{2}$$

2. In this systems of unit

$$\hbar = 1, \quad e^2 = 2, \quad m = \frac{1}{2}$$

3. For spin-orbit coupling or anything else, we need to have numbers for  $c$ .

$$\frac{e^2}{\hbar c} \approx \frac{1}{137}, \quad c \approx 274$$

# Atomic-Rydberg systems of unit

$$\left( -\nabla^2 + U_{atom}(\vec{r}) + \int \frac{2\rho(\vec{r}')}{|\vec{r} - \vec{r}'|} d^3\vec{r}' + v_{xc}[\rho(\vec{r})] \right) \psi_n(\vec{r}) = \varepsilon_n \psi_n(\vec{r})$$

# Atomic-Hatree systems of unit

1. Atomic Rydberg systems of unit can be derived by setting

$$a_0 = \frac{\hbar^2}{me^2} = 1, \quad R = \frac{e^2}{2a_0} = \frac{me^4}{2\hbar^2} = 1$$

2. In this systems of unit

$$\hbar = 1, \quad e^2 = 1, \quad m = 1$$

3. For spin-orbit coupling or anything else, we need to have numbers for  $c$ .

$$\frac{e^2}{\hbar c} \approx \frac{1}{137}, \quad c \approx 137$$